

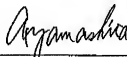
# DECLARATION

I, Atsushi YAMASHITA of UNIUS PATENT ATTORNEYS OFFICE hereby declare as follows:

1. I am a translator and am familiar with both English and Japanese languages.

2. I am the translator of the documents attached hereto and certify that the following is a true translation of Japanese Patent Application No. 2003-122951 to the best of my knowledge and belief.

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【Title of the invention】 REACTIVE MODIFIER

【Claims】

[Claim 1] A reactive modifier which is an acrylic polymer having a number average molecular weight of 5000 or more, obtained by copolymerizing (A) a hydrolyzable silyl group-containing monomer (B) methyl methacrylate, (C) butyl acrylate and (D) an alkyl (meth) acrylate in which the alkyl chain has 7 to 9 carbon atoms, wherein:

the content of the component A is 0.01 to 10 parts by weight, the content of the component B is 5 to 95 parts by weight, the content of the component C is 5 to 95 parts by weight, the content of the component D is 5 to 95 parts by weight, the total content of the components A to D is 50 to 100 parts by weight and the weight ratio of the component C to the component D is 0.5 to 2.0.

[Claim 2] The reactive modifier according to claim 1, which is a modifier for an oxyalkylene polymer having silicon-containing functional groups capable of cross-linking by forming siloxane bonds.

[Claim 3] A room temperature-curable resin composition which comprises an oxyalkylene polymer having silicon-containing functional groups capable of cross-linking by forming siloxane bonds, the reactive modifier according to claim 1 and a curing accelerator.

[The detailed description of the invention]

[0001]

[Technical field of the invention]

This invention relates to a reactive modifier for a curable composition, particularly, for an oxyalkylene polymer which contains hydrolyzable silyl groups and to a room temperature-curable composition, including the modifier, excellent in storage stability. In the present invention, an alkyl (meth) acrylate means an alkyl acrylate and/or an alkyl methacrylate.

[0002]

[Conventional Art]

Methods in which reactive silicon functional group-containing (meth)acrylate polymers are used as modifiers for reactive silicon functional group-containing oxyalkylene polymers are proposed in Patent Documents 1 to 4.

Among these methods, the method disclosed in Patent Document 3 in which a copolymer containing a long-chain alkyl (meth)acrylate is blended is excellent from the viewpoints of transparency, tensile strength and other various practical properties. However, in the method in which a copolymer containing a long-chain alkyl (meth)acrylate is blended, there are problems that a long-chain alkyl (meth)acrylate monomer as raw material is generally expensive and is not easy to handle because it is solid at normal temperature. Also, when two or more alkyl (meth)acrylates are copolymerized, it is generally necessary to cool the monomer units mixed together in advance for the sake of safety; however, in the case where a long-chain alkyl (meth)acrylate monomer is among

the monomers used, there is a problem that it is unfavorably segregated when mixed with the cooled other monomers.

[0003]

[Patent Document 1] Japanese Patent Laid-Open No. 59-122541

[0004]

[Patent Document 2] Japanese Patent Laid-Open No. 60-31556

[0005]

[Patent Document 3] Japanese Patent Laid-Open No. 63-112642

[0006]

[Patent Document 4] Japanese Patent Laid-Open No. 6-172631

[0007]

[The subject to be solved by the invention]

An object of the present invention is to develop a reactive modifier which is inexpensive and favorable in handling of the raw materials thereof, and can provide a curable composition capable of exhibiting excellent storage stability and excellent mechanical properties.

[0008]

[Means solving the subject]

The present inventors have achieved the present invention by finding that even when, in a reactive silicon group-containing alkyl (meth)acrylate polymer, an alkyl (meth)acrylate containing an alkyl group having 7 to 9 carbon atoms is used in a specific amount as a copolymerization component, instead of a long-chain an alkyl (meth)acrylate, there is obtained a reactive modifier for a reactive silicon group-containing resin having a sufficient storage stability and being excellent in various practical properties.

[0009]

The present invention relates to a reactive modifier which is an acrylic polymer having a number average molecular weight of 5000 or more, obtained by copolymerizing (A) a hydrolyzable silyl group-containing monomer (B) methyl methacrylate, (C) butyl acrylate and (D) an alkyl (meth) acrylate in which the alkyl chain has 7 to 9 carbon atoms.

[0010]

The reactive modifier of the present invention is characterized in that the content of the component B is 10 to 55 parts by weight.

[0011]

The reactive modifier of the present invention is characterized in that the weight ratio of the component C to the component D is 0.5 to 2.0.

[0012]

Moreover, the present invention relates to a room temperature-curable resin composition which comprises an oxyalkylene polymer having silicon-containing functional groups capable of cross-linking by forming siloxane bonds, the reactive modifier and a curing accelerator.

[0013]

[Embodiment of the invention]

The present invention will be described below in detail.

[0014]

The reactive silicon group in the hydrolyzable silyl group-containing monomer as the component (A) in the present invention is characterized in that it is capable of cross-linking even at room temperature by forming siloxane bonds. A typical example of the reactive silicon group is represented by the general formula (1):



wherein R<sup>1</sup> represents an alkyl group having 1 to 20 carbon atoms, an aryl group having 6 to 20 carbon atoms, an aralkyl group having 7 to 20 carbon atoms, or a triorganosiloxy group represented by (R')<sub>3</sub>SiO-; wherein R' is a monovalent hydrocarbon group having 1 to 20 carbon atoms, the three R's may be the same or different; X represents a hydroxy group or a hydrolyzable group, and when two or more Xs are present, they may be the same or different; and a represents 0, 1, 2 or 3.

[0015]

From the viewpoints of economic efficiency, handling easiness and the like, the specific example of the reactive silicon group in the component (A) is preferably at least one selected from the group consisting of a dimethylmonomethoxysilyl group, a methyl dimethoxysilyl group, a trimethoxysilyl group, an ethyldiethoxysilyl group, a triethoxysilyl group, a methyl diisopropenyloxysilyl group and a triisopropenyloxysilyl group.

[0016]

The alkyl (meth)acrylate monomer, as the component (D) in the present invention, containing an alkyl group having 7 to 9 carbon atoms is represented by the general formula (2):



wherein R<sup>2</sup> represents a hydrogen atom or a methyl group, and R<sup>3</sup> represents an alkyl group having 7 to 9 carbon atoms. Examples of R<sup>3</sup> may include alkyl groups such as an n-heptyl group, an isohexyl group, an n-octyl group, a 2-ethylhexyl group, an isooctyl group, an n-nonyl group and an isononyl group. However, a 2-ethylhexyl group and an isononyl group are particularly preferable from the viewpoints of the availability of the raw materials and the economic efficiency. The monomers represented

by the general formula (2) may be used as the component (D) each alone or in combinations of two or more thereof.

[0017]

For the purpose of attaining sufficient performance as a reactive modifier, it is necessary that the reactive modifier be sufficiently compatible with the polymer to be modified and the viscosity of the mixture composed of the reactive modifier and the polymer to be modified be appropriate.

[0018]

Methyl methacrylate as the component (B) and butyl acrylate as the component (C) in the present invention are relatively inexpensive and easily available monomers, which are copolymerized in combination with the component (D) to permit attaining a sufficient degree of polymerization. It is also possible to adjust the T<sub>g</sub> of the reactive modifier in such a way that the T<sub>g</sub> meets the purpose of ensuring the sufficient compatibility of the reactive modifier with various polymers and the purpose of obtaining appropriate viscosities of the mixtures of the reactive modifier with various polymers.

[0019]

The number of the reactive silicon group in one molecule of the reactive modifier in the present invention is, on average, preferably one or more, more preferably 1.1 or more, and particularly preferably 1.5 or more because it is desirable that a sufficient curability is attained and the reactive modifier, for the purpose of displaying sufficient performance as reactive modifier, undergoes co-cross-linking with the polymer to be modified.

[0020]



The reactive modifier having a number average molecular weight of 5,000 to 100,000, as measured by the GPC method relative to polystyrene standards, is preferable from the viewpoints of the handling easiness, the possibility of obtaining a sufficient compatibility with the polymer to be modified and the possibility of reducing the used amount of the expensive reactive silicon group group-containing monomer. The number average molecular weight of the reactive modifier is more preferably 5,000 to 50,000, and particularly preferably 7,000 to 30,000. When the number average molecular weight of the reactive modifier is less than 5,000, a satisfactory rubber elastic substance is hardly obtainable, and a disadvantage in cost is caused by the inevitable increase of the used amount of the reactive silicon group-containing monomer. When the number average molecular weight of the reactive modifier exceeds 100,000, there is a fear that its viscosity becomes high to make the handling thereof difficult.

**[0021]**

The reactive modifier of the present invention is obtained by vinyl polymerization, for example, a radical reaction-based vinyl polymerization of the monomers (A), (B), (C) and (D), and more specifically, by usual polymerization methods such as a solution polymerization method and a mass polymerization method. The polymerization concerned can be carried out by adding a radical initiator to these monomers according to need at a reaction temperature of 50 to 150°C. A solvent may be used or may not be used. When a solvent is used, it is preferable to use a non-reactive solvent selected from the solvents such as ethers, hydrocarbons, acetates and alcohols from the viewpoints that these solvents are inexpensive and meet the safety of the polymerization reaction. From an environmental viewpoint, it is preferable to use

non-aromatic solvents. As non-aromatic solvents, alcohols such as butanol lead to preferable results from the viewpoints of environmental effects and the handling easiness of the obtained polymers.

【0022】

Various methods are available as the method for introducing the hydrolyzable silyl groups into the reactive modifier in the present invention. Examples of such methods may include the following methods.

(1) A method in which a compound (for example,  $\text{CH}_2=\text{CHSi}(\text{OCH}_3)_3$ ) having a polymerizable unsaturated bond and a reactive silicon group is added to a mixture of the monomers (B), (C) and (D) to be copolymerized with each other.

(2) A method in which a compound (for example, acrylic acid) having a polymerizable unsaturated bond and a reactive functional group (hereinafter referred to as a Z group) is added to a mixture of the monomers (B), (C) and (D) to be copolymerized with each other, and thereafter the produced copolymer is reacted with a compound (for example, a compound having an isocyanate group and a  $-\text{Si}(\text{OCH}_3)_3$  group) having a reactive silicon group and a functional group (hereinafter referred to as a Z' group) capable of reacting with the Z group. (3) The monomers (B), (C) and (D) are copolymerized with each other by living radical polymerization to introduce reactive silicon group into the terminals of the molecular chain however, above mentioned method is not limited to these examples.

【0023】

Specific examples of the compound used in the above described method

(1) may include:  $\text{CH}_2=\text{CHSi}(\text{CH}_3)(\text{OCH}_3)_2$ ,  $\text{CH}_2=\text{CHSi}(\text{CH}_3)\text{Cl}_2$ ,  $\text{CH}_2=\text{CHSi}(\text{OCH}_3)_3$ ,  $\text{CH}_2=\text{CHSiCl}_3$ ,  $\text{CH}_2=\text{CHCOO}(\text{CH}_2)_2\text{Si}(\text{CH}_3)(\text{OCH}_3)_2$ ,  $\text{CH}_2=\text{CHCOO}(\text{CH}_2)_2\text{Si}(\text{OCH}_3)_3$ ,  $\text{CH}_2=\text{CHCOO}(\text{CH}_2)_2\text{Si}(\text{CH}_3)\text{Cl}_2$ ,  $\text{CH}_2=\text{CHCOO}(\text{CH}_2)_2\text{SiCl}_3$ ,  $\text{CH}_2=\text{C}(\text{CH}_3)\text{COO}(\text{CH}_2)_2\text{Si}(\text{CH}_3)(\text{OCH}_3)_2$ ,  $\text{CH}_2=\text{C}(\text{CH}_3)\text{COO}(\text{CH}_2)_2\text{Si}(\text{OCH}_3)_3$ ,

$\text{CH}_2=\text{C}(\text{CH}_3)\text{COO}(\text{CH}_2)_3\text{Si}(\text{CH}_3)(\text{OCH}_3)_2$ ,  $\text{CH}_2=\text{C}(\text{CH}_3)\text{COO}(\text{CH}_2)_3\text{Si}(\text{OCH}_3)_3$ ,  
 $\text{CH}_2=\text{C}(\text{CH}_3)\text{COO}(\text{CH}_2)_2\text{Si}(\text{CH}_3)\text{Cl}_2$ ,  $\text{CH}_2=\text{C}(\text{CH}_3)\text{COO}(\text{CH}_2)_2\text{SiCl}_3$ ,  
 $\text{CH}_2=\text{CHCH}_2\text{OC}(\text{O})-\text{Ph}-\text{COO}(\text{CH}_2)_3\text{Si}(\text{CH}_3)(\text{OCH}_3)_2$ ,  
 $\text{CH}_2=\text{CHCH}_2\text{OC}(\text{O})-\text{Ph}-\text{COO}(\text{CH}_2)_3\text{Si}(\text{OCH}_3)_3$ ,  
 $\text{CH}_2=\text{CHCH}_2\text{OC}(\text{O})-\text{Ph}-\text{COO}(\text{CH}_2)_3\text{Si}(\text{CH}_3)\text{Cl}_2$ , and  
 $\text{CH}_2=\text{CHCH}_2\text{OC}(\text{O})-\text{Ph}-\text{COO}(\text{CH}_2)_3\text{SiCl}_3$ . Herein, Ph represents a p-phenylene group. Among these compounds,  $\text{CH}_2=\text{CHSi}(\text{OCH}_3)_3$ ,  
 $\text{CH}_2=\text{C}(\text{CH}_3)\text{COO}(\text{CH}_2)_3\text{Si}(\text{CH}_3)(\text{OCH}_3)_2$  and  $\text{CH}_2=\text{C}(\text{CH}_3)\text{COO}(\text{CH}_2)_3\text{Si}(\text{OCH}_3)_3$  are preferable from the viewpoints of the economic efficiency and the reactivity of the curable composition to be obtained.

[0024]

These silane compounds are synthesized by various methods, such as a method in which acetylene, allyl acrylate, allyl methacrylate, diallyl phthalate or the like is reacted with methyldimethoxysilane, methyl dichlorosilane or the like in the presence of group VIII transition metal catalysts. As such transition metal complex catalysts, those complex compounds of the group VIII transition metals selected from the group consisting of platinum, rhodium, cobalt, palladium and nickel are effectively used. Particularly effective are platinum black and platinum compounds such as chloroplatinic acid, platinum-alcohol compounds, platinum-olefin complexes, platinum-aldehyde complexes and platinum-ketone complexes.

[0025]

As for the compounds used in the method (2), examples of the Z group and the Z' group may include various combinations of groups, among which is an example in which the Z group is a vinyl group and the Z' group is a hydrosilicon group (H-Si). The Z group and the Z' group can be combined through hydrosilylation reaction. Examples of the compound

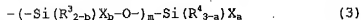
which has a vinyl group as the Z group and further has a polymerizable unsaturated bond may include allyl acrylate and allyl methacrylate. Specific examples of the hydrosilane compound which is a compound having a hydrosilicon group as the Z' group and further having a reactive silicon group may include: alkoxysilanes such as trimethoxysilane, triethoxysilane, methyldiethoxysilane, methyldimethoxysilane, dimethylmethoxysilane and phenyldimethoxysilane; acyloxysilanes such as methyldiacetoxysilane and trimethylsiloxyethylacetoxysilane; ketoximatesilanes such as bis(dimethylketoximate)methylsilane, bis(cyclohexylketoximate)methylsilane and bis(diethylketoximate)trimethylsiloxyethylsilane; hydrosilanes such as dimethylsilane, trimethylsiloxyethylsilane and 1,1-dimethyl-2,2-dimethyldisiloxane; and alkenyloxysilanes such as methyltri(isopropenyloxy)silane and dimethyltri(isopropenyloxy)silane. However, the above mentioned hydrosilane compound is not limited to these examples.

[0026]

Example of the method disclosed (3) may include those disclosed in Japanese Patent Laid-open No. 09-272714.

[0027]

No particular constraint is imposed on the reactive silicon group groups contained in the silicon-containing functional group-containing oxyalkylene polymer used in the present invention; however, typical examples of such hydrolyzable silyl groups may include the groups represented by the general formula (3):



wherein  $\text{R}^3$  and  $\text{R}^4$  each represent an alkyl group having 1 to 20 carbon atoms, an aryl group having 6 to 20 carbon atoms, an aralkyl group having

7 to 20 carbon atoms or a triorganosiloxy group represented by  $(R')_3SiO-$ ; when two or more  $R^1$ 's or  $R^2$ 's are present, they may be the same or different; wherein  $R'$  is a monovalent hydrocarbon group having 1 to 20 carbon atoms, and 3  $R'$ 's may be the same or different; X represents a hydroxy group or a hydrolyzable group, and when two or more Xs are present, they may be the same or different; a represents 0, 1, 2 or 3; b represents 0, 1 or 2; and the c-values in the m expressions each represented by the general formula (3) may be the same or different:



and m represents an integer of 0 to 19, with the proviso that  $b + \sum c \geq 1$  is satisfied.

No particular constraint is imposed on the hydrolyzable group represented by the above described X, and any hydrolyzable groups well known in the art may be suitably used. Specific examples of the hydrolyzable group may include: a hydrogen atom, a halogen atom, an alkoxy group, an acyloxy group, a ketoximate group, an amino group, an amide group, an acid amide group, an aminooxy group, a mercapto group and an alkenyloxy group. Among these, a hydrogen atom, an alkoxy group, an acyloxy group, a ketoximate group, an amino group, an amide group, an aminooxy group, a mercapto group and an alkenyloxy group are preferable from the viewpoints of the environmental consideration and easy availability of the raw materials involved, alkoxy groups such as a methoxy group being particularly preferable because of the moderate hydrolyzability and handling easiness.

[0028]

To one silicon atom, 1 to 3 hydrolyzable and hydroxy groups can be bonded, and  $(b + \sum c)$  preferably falls within a range from 1 to 5. When two or more hydrolyzable and / or hydroxy groups are present in

a reactive silicon, those groups may be the same or different. In the reactive silicon, one silicon atom may be present, or two or more silicon atoms may be present; in the case of a hydrolyzable silyl group having silicon atoms combined with each other through siloxane bonds or the like, about 20 silicon atoms may be present. No particular constraint is imposed on the hydrolyzable silicon group; however, in consideration of the balance between the high hydrolysis activity and the moderate hydrolyzability combined with the handling easiness, the hydrolyzable silicon group is preferably at least one selected from the group consisting of a dimethylmonomethoxysilyl group, a methyl dimethoxysilyl group, a trimethoxysilyl group, an ethyldiethoxysilyl group, a triethoxysilyl group, a methyl diisopropenyloxysilyl group and a triisopropenyloxysilyl group.

[0029]

In one molecule of the oxyalkylene polymer, preferably at least one reactive silicon group is present, and more preferably about 1.1 to 5 the reactive silicon groups are present. When the number of the hydrolyzable silyl groups contained in one molecule of the polymer is less than one, unpreferably the curability becomes insufficient and a satisfactory rubber elasticity is hardly attainable. When the number of the hydrolyzable silyl groups exceeds 5, unpreferably the cured articles become too hard. The hydrolyzable silyl groups may be present at the terminals or in an interior portion of the molecular chain of the oxyalkylene polymer. When the hydrolyzable silyl groups are present at the terminals of the molecular chain, the content of the effective network chain in the oxyalkylene polymer component contained in a finally formed cured article is increased, so that it preferably becomes easier

to obtain a rubber-like cured article exhibiting a high strength, a high elongation and a low elastic modulus.

[0030]

Examples of the oxyalkylene polymer may include oxyethylene, oxypropylene, oxybutylene, oxyisobutylene and oxytetramethylene. However, the oxyalkylene is not limited to these examples. The molecular chain of the oxyalkylene polymer may be formed of one type of repeating unit or two or more types of repeating units. The oxyalkylene polymer may also be a straight chain, a branched chain or an admixture of straight chains and branched chains. Among these oxyalkylenes, preferable is the one which has a molecular chain substantially including a repeating unit represented by the general formula (5):



from the viewpoints of the handling easiness of the obtained curable composition and the physical properties of the cured articles. The term "substantially" as referred to herein means that the repeating unit represented by the general formula (5) is present in the polymer in a content of 50 wt% or more, preferably 80 wt% or more, wherein other monomers or the like may be present.

[0031]

The number average molecular weight ( $M_n$ ) of the oxyalkylene polymer is preferably 6,000 to 60,000 as measured by the gel permeation chromatography (GPC) method relative to polystyrene standards, more preferably 8,000 to 50,000 from the viewpoints of the curability and handling easiness, and particularly preferably 10,000 to 30,000 from the viewpoint of the excellent mechanical properties. The ratio of the weight average molecular weight to the number average molecular weight, ( $M_w/M_n$ ), is preferably as narrow in molecular weight distribution (small

in the ratio  $M_w/M_n$ ) as 1.6 or less, more preferably 1.5 or less, and particularly preferably 1.4 or less. When the oxyalkylene polymer (a) having such a narrow molecular weight distribution is mixed with the reactive silicon group-containing alkylester (meth) acrylate monomer (b), the curable composition preferably has a low viscosity to attain a satisfactory workability as compared to the case where an oxyalkylene polymer (a) having a wide molecular weight distribution is used.

[0032]

The reactive silicon group-containing oxyalkylene polymer of the present invention may be obtained, for example, by introducing reactive silicon groups into an oxyalkylene polymer having functional groups.

[0033]

The introduction of the reactive silicon groups may be carried out on the basis of methods well known in the art. More specifically, the following methods may be cited.

[0034]

<1> An oxyalkylene polymer having functional groups such as hydroxy groups at the terminals thereof is reacted with an organic compound having an active group capable of reacting with the functional groups and an unsaturated group, and then the obtained reaction product is reacted with a hydrosilane having a hydrolyzable group to carry out hydrosilylation.

[0035]

<2> An oxyalkylene polymer having functional groups (hereinafter referred to as functional groups Y) such as hydroxy group, an epoxy group and an isocyanate group at the terminals thereof is reacted with a compound having a functional group (hereinafter referred to as a functional group Y') capable of reacting with the functional groups Y and also having



a reactive silicon group. Specific examples of the silicon compound having the functional group Y' may include: aminogroup-containing silanes such as  $\gamma$ -(2-aminoethyl)aminopropyltrimethoxysilane,  $\gamma$ -(2-aminoethyl)aminopropylmethyldimethoxysilane and  $\gamma$ -aminopropyltriethoxysilane; mercapto group-containing silanes such as  $\gamma$ -mercaptopropyltrimethoxysilane and  $\gamma$ -mercaptopropylmethyldimethoxysilane; epoxy silanes such as  $\gamma$ -glycidoxypropyltrimethoxysilane and  $\beta$ -(3,4-epoxycyclohexyl)ethyltrimethoxysilane; vinylic unsaturated group-containing silane groups such as vinyltriethoxysilane,  $\gamma$ -methacryloyloxypropyltrimethoxysilane and  $\gamma$ -acryloyloxypropylmethyldimethoxysilane; chlorine atom-containing silanes such as  $\gamma$ -chloropropyltrimethoxysilane; isocyanate-containing silanes such as  $\gamma$ -isocyanatepropyltriethoxysilane and  $\gamma$ -isocyanatepropylmethyldimethoxysilane; and hydrosilanes such as methyldimethoxysilane, trimethoxysilane and methyldiethoxysilane. However, the silicon compound having the functional group Y' is not limited to these examples.

[0036]

In the above methods, preferable is the method <1> or <2> in which a polymer having hydroxy groups at the terminals thereof is reacted with a compound having an isocyanate group and a the reactive silicon group, from the viewpoints of the reaction easiness and the availability of the raw materials.

[0037]

Because the molecular weight distribution of a polymer tends to be broadened when reactive silicon groups are introduced as compared to the molecular weight distribution of the polymer before introduction

of such silyl groups, it is preferable that the molecular weight distribution of the polymer before the introduction of the hydrolyzable silyl groups is as narrow as possible.

[0038]

Oxyalkylene polymers each having a high molecular weight and a narrow molecular weight distribution may be obtained, for example, by the methods disclosed in Japanese Patent Laid-Open Nos. 61-197631, 61-215622, 61-215623 and 61-218632; Japanese Patent Publication Nos. 46-27250 and 59-15336; and Japanese Patent Laid-Open Nos. 50-149797, 61-197631, 2-276821, 10-273512, 10-36499, 11-106500, 11-302371 and the like.

[0039]

Among these, for the purpose of easily obtaining a polymer having a narrow molecular weight distribution, preferable is a polymer which is obtained by polymerizing an alkylene oxide in the presence of an initiator by use of a compound selected from the group consisting of a composite metal cyanide complex, a cesium compound and a P=N bond-containing compound as a catalyst.

[0040]

As the initiator, usable are dihydric alcohols and polyhydric alcohols such as ethylene glycol, propylene glycol, butanediol, hexamethylene glycol, methallyl alcohol, hydrogenated bisphenol A, neopentyl glycol, polybutadiene diol, diethylene glycol, triethylene glycol, polyethylene glycol, dipropylene glycol, glycerin, trimethylolmethane, trimethylolpropane, pentaerythritol and the like; and various oligomers having hydroxy groups. Among these, preferable are polypropylene glycol, polypropylene triol and polypropylene tetraol from the viewpoints of the economic efficiency and handling easiness.

[0041]

As the composite metal cyanide complex usable in the present invention, complexes containing zinc hexacyanocobaltate as main component are preferable from the viewpoint of the polymerization activity, among which the ether and/or alcohol complexes are preferable from the viewpoint of the polymerization control. For the purpose of obtaining a polymer having a narrower molecular weight distribution, ethyleneglycol dimethyl ether (glyme) and diethylene glycol dimethyl ether (diglyme) are preferable as ethers, and t-butanol is preferable as an alcohol.

[0042]

The used amount of a composite metal cyanide complex is preferably 0.0001 to 0.03 wt% in terms of the content thereof in the final polyoxyalkylene compound, and more preferably 0.001 to 0.01 wt% from the viewpoint of the reactivity. When the content is less than 0.0001 wt%, unpreferably no sufficient reaction rate is attained, while when the content exceeds 0.03 wt%, unpreferably the production cost of the oxyalkylene polymer is increased.

[0043]

As the cesium compound usable in the present invention, from the viewpoint of the polymerization reactivity, preferable is a compound having as main component an item selected from the group consisting of metallic cesium; cesium alkoxide such as cesium methoxide, cesium ethoxide and cesium propoxide; and cesium hydroxide and cesium carbonate. Cesium hydroxide is more preferable from the viewpoints of the availability and the economic efficiency.

[0044]

The used amount of a cesium-based catalyst is preferably 0.05 to 1.5 wt% in terms of the content of metallic cesium in the final

polyoxyalkylene compound, and from the viewpoint of the reactivity, more preferably 0.1 to 1.0 wt%. When the content is less than 0.05 wt%, unpreferably no sufficient reaction rate is attained, while when the content exceeds 1.5 wt%, unpreferably the production cost of the oxyalkylene polymer is increased.

[0045]

A preferable form of the P=N bond-containing compound usable in the present invention may include at least one compound selected from the group consisting of phosphazanium compounds, phosphineoxide compounds and phosphazene compounds.

[0046]

As phosphazanium compounds, there may be cited the compounds described in Japanese Patent Laid-Open No. 11-106500. Examples of such compounds may include:

tetrakis[tris(dimethylamino)phosphoranylidenamino]phosphonium hydroxide,

tetrakis[tris(dimethylamino)phosphoranylidenamino]phosphonium methoxide,

tetrakis[tris(dimethylamino)phosphoranylidenamino]phosphonium ethoxide and

tetrakis[tri(pyrrolidin-1-yl)phosphoranylidenamino]phosphonium tert-butoxide.

[0047]

As phosphazene compounds, there may be cited the compounds described in Japanese Patent Laid-Open No. 10-36499. Examples of such compounds may include: 1-tert-butyl-2,2,2-tris(dimethylamino)phosphazene, 1-(1,1,3,3-tetramethylbutyl)-2,2,2-tris(dimethylamino)phosphazene,

1-ethyl-2,2,4,4,4-pentakis(dimethylamino)-2λ5,4λ5-catenadi(phosphazene),

1-tert-butyl-4,4,4-tris(dimethylamino)-2,2-bis[tris(dimethylamino)phosphoranylidenamino]-2λ5,4λ5-catenadi(phosphazene),

1-(1,1,3,3-tetramethylbutyl)-4,4,4-tris(dimethylamino)-2,2-bis[tris(dimethylamino)phosphoranylidenamino]-2λ5,4λ5-catenadi(phosphazene),

1-tert-butyl-2,2,2-tri(1-pyrrolidinyl)phosphazene and

7-ethyl-5,11-dimethyl-1,5,7,11-tetraaza-6λ5-phosphaspiro[5,5]undecan-1(6)-ene.

[0048]

As phosphine oxide compounds, there may be cited the compounds described in Japanese Patent Laid-Open No. 11-302371. Examples of such compounds may include:

tris[tris(dimethylamino)phosphoranylidenamino]phosphine oxide and tris[tris(diethylamino)phosphoranylidenamino]phosphine oxide.

[0049]

Among these, from the viewpoints of industrial applications, phosphazanium compounds and phosphine oxide compounds are preferable.

[0050]

The used amount of a P=N bond-containing compound is preferably such that the P=N bond-containing compound is contained in a content of  $1 \times 10^{-4}$  to  $5 \times 10^{-1}$  mol in relation to 1 mol of the active hydrogen compound in the initiator. When the content is less than  $1 \times 10^{-4}$  mol, unpreferably no sufficient reaction rate is attained, while when the content exceeds  $5 \times 10^{-1}$  mol, unpreferably the production cost of the polyoxyalkylene compound is increased.

[0051]

As for the proportion between the reactive modifier and the oxyalkylene polymer in the curable composition of the present invention, the amount of the reactive modifier is set preferably within a range from 5 to 5000 parts (by weight, and the same hereinafter) in relation to 100 parts of the oxyalkylene polymer because such a range provides a remarkable effect in improving the properties of the oxyalkylene polymer, and more preferably within a range from 5 to 2000 parts, the amount of the reactive modifier being usually selected according to the intended applications and performances.

[0052]

In the composition of the present invention, a curing accelerator may be used, no particular constraint being imposed on the type of the curing accelerator. However, specific examples of the curing catalysts may include tin catalysts, non-tin metal catalysts and basic compounds. Examples of the tin catalysts may include: divalent tin carboxylates such as tin octylate, tin oleate and tin stearate; dibutyltin dicarboxylates such as dibutyltin dilaurate and dibutyltin bis(alkylmaleate); dialkyltin alkoxide derivatives such as dibutyltin dimethoxide and dibutyltin diphenoxide; intramolecular coordinative derivatives of dialkyltin such as dibutyltin diacetylacetonate and dibutyltin acetoacetate; reaction mixtures between dibutyltin oxide and ester compounds, reaction mixtures between dibutyltin oxide and silicate compounds, and derivatives of tetravalent dialkyltin oxides such as oxy derivatives of these dialkyltin oxide derivatives. Examples of the non-tin metal catalysts may include: metal carboxylates containing, as carboxylic acid components, octylic acid, oleic acid, naphthenic acid or stearic acid, such as calcium carboxylates, zirconium carboxylates, iron carboxylates, vanadium carboxylates, bismuth carboxylates, lead

carboxylates, titanium carboxylates and nickel carboxylates; titanium alkoxides such as tetraisopropyl titanate, tetrabutyl titanate, tetramethyl titanate and tetra(2-ethylhexyl titanate); aluminum alkoxides such as aluminum isopropylate, mono-sec-butoxy aluminum diisopropylate and aluminum sec-butylate; zirconium alkoxides such as zirconium n-propylate and zirconium n-butylate; titanium chelates such as titanium tetraacetylacetonate, titanium ethylacetoacetate, octylene glycolate and titanium lactate; aluminum chelates such as aluminum trisacetylacetonate, aluminum trisethylacetoacetate, diisopropoxy aluminium ethylacetoacetate; and zirconium chelates such as zirconium tetraacetylacetonate, zirconium monoacetylacetonate, zirconium bisacetylacetonate, zirconium acetylacetonate bisethylacetoacetate and zirconium acetate. Examples of the basic compounds may include: amines, amine salts, quaternary ammonium salts and guanidine compounds.

[0053]

These may be used each alone or as admixtures thereof. These curing catalysts are used in an amount of about 0.1 to 10 parts by weight in relation to 100 parts by weight of the the reactive silicon group-containing oxyalkylene polymer(A). When the used amount of the curing accelerator(s) is too small, sometimes the curing rate becomes slow, and in some cases, the curing reaction hardly tends to proceed to a sufficient extent. On the other hand, when the used amount of the curing accelerator(s) is too large, local heat evolution or local foaming unpreferably occurs at the time of curing to make it difficult to obtain satisfactory cured articles.

[0054]

When the curable composition of the present invention is used, there may be optionally used fillers including reinforcing fillers such as

fumed silica, precipitated silica, silicic anhydride, hydrous silicic acid and carbon black; fillers such as calcium carbonate, magnesium carbonate, diatomite, calcined clay, clay, talc, titanium oxide, bentonite, organic bentonite, ferric oxide, zinc oxide, active zinc white, hydrogenated castor oil and shirasu balloon; and fibrous fillers such as asbestos, glass fiber and glass filament. When a curable composition having a high strength is desired to be obtained, preferable results are obtained when a filler selected from the group consisting of fumed silica, precipitated silica, silicic anhydride, hydrous silic acid, carbon black, surface treated fine calcium carbonate, calcined clay, clay and active zinc white and the like is mainly used within a range from 1 to 100 parts by weight in relation to 100 parts by weight of the hydrolyzable silyl group-containing oxypropylene polymer. Also, when a curable composition having a low strength and a large elongation is desired to be obtained, preferable results are obtained when a filler selected from the group consisting of titanium oxide, calcium carbonate, magnesium carbonate, talc, ferric oxide, zinc oxide and shirasu balloon and the like is mainly used within a range from 5 to 200 parts by weight in relation to 100 parts by weight of the reactive silicon group-containing oxypropylene polymer. Needless to say, these fillers may be used each alone or as admixtures of two or more thereof.

[0055]

The use of a plasticizer, in combination with a filler, in the curable composition of the present invention is more effective because the elongation of the cured articles is made thereby larger and the amount of a filler or fillers mixed in the curable composition can be thereby made larger. As such a plasticizer, the following compounds may be optionally used each alone or as admixtures of two or more thereof:



phthalates such as dioctyl phthalate, dibutyl phthalate and butyl benzyl phthalate; aliphatic dibasic acid esters such as dioctyl adipate, isodecyl succinate and dibutyl sebacate; glycol esters such as diethylene glycol dibenzoate and pentaerythritol ester; aliphatic esters such as butyl oleate and methyl acetylricinoleate; phosphates such as tricresyl phosphate, trioctyl phosphate and octyl diphenyl phosphate; epoxy plasticizers such as epoxidized soy-bean oil, epoxidized flaxseed oil and benzyl epoxystearate; polyester plasticizers such as polyesters between dibasic acids and dihydric alcohols; polyethers such as polypropylene glycol and derivatives thereof; polystyrenes such as poly- $\alpha$ -methylstyrene and polystyrene; and polybutadiene, butadiene/acrylonitrile copolymer, polychloroprene, polyisoprene, polybutene and chlorinated paraffins. Preferable results are obtained when the amount of a plasticizer or plasticizers used falls within a range from 0 to 100 parts by weight in relation to 100 parts by weight of the reactive silicon group-containing oxypropylene polymer.

[0056]

According to need, there may also be optionally added various additives such as adhesion improvers, physical property adjusters, storage stability improvers, antiaging agents, ultraviolet absorbers, metal deactivators, antiozonants, photostabilizers, amine-based radical chain inhibitors, phosphorus-based peroxide decomposers, lubricants, pigments and foaming agents. No particular constraint is imposed on the preparation method of the curable composition of the present invention; thus, there may be adopted common methods in which, for example, the above described ingredients are compounded and the mixture thus obtained is kneaded at room temperature or under heating with a mixer, roll, kneader or the like, or the mixture is dissolved by use of a small

amount of an appropriate solvent to mix the ingredients therein. Appropriate combinations of these ingredients permit preparing a one-component composition or a two-component composition to be applied.

[0057]

The curable composition of the present invention forms three dimensional network structure by the action of moisture when exposed to the atmosphere, to be cured into a solid article having rubber-like elasticity. The curable composition of the present invention is particularly useful as an elastic sealant, and may be used as sealants for buildings, ships, vehicles, roads and the like. Also, the curable composition concerned can adhere, alone or with the aid of primers, to a wide range of substrates such as glass, porcelain, wood, metals and molded resin articles, so that it can be used as various types of sealing compositions and adhesive compositions. It can be used as adhesives such as one-component adhesives, two-component adhesives, contact adhesives to adhere after an open time and tackifiers; it is also useful as coating materials, waterproof coated films, food packaging materials, cast rubber materials, molding materials and foaming materials.

[0058]

[Example]

For the purpose of more clearly describing the present invention, description will be made below with reference to specific examples, but the present invention is not limited to these examples. Here, it may be noted that the hydrolyzable silyl group-containing monomers (A) are as follows:

Y9910:  $\gamma$ -Methacryloxypropylmethyldimethoxysilane  
(manufactured by Japan Unicar Co., Ltd.)

A-174:  $\gamma$ -Methacryloxypropyltrimethoxysilane

(Preparation Example 1)

Propylene oxide was polymerized by use of a polyoxypropylene glycol having a molecular weight of about 2000 as initiator and a zinc hexacyanocobaltate-glyme complex catalyst, to yield a polyoxypropylene glycol (namely, a hydroxy group-terminated polyether oligomer) having an average molecular weight of 10,000 as measured by a terminal group analysis. Then, there was added a methanol solution of 1.2 equivalents of NaOMe in relation to the hydroxy groups of this hydroxy group-terminated polyether oligomer, and then the methanol was distilled off. There was further added 3-chloro-1-propene to convert the terminal hydroxy groups into allyl groups. Then, 10 g of hexane was added to 500 g of the obtained oligomer, and azeotropic dehydration was carried out at 90°C, the hexane was distilled off under reduced pressure, and purging with nitrogen was effected. To the thus treated oligomer, 30  $\mu$ l of platinum-divinylidisiloxane complex (an isopropanol solution of 3 wt% in terms of platinum) was added, and to the reaction mixture thus obtained, 9.0 g of DMS (dimethoxymethylsilane) was slowly added dropwise under stirring. After allowing the mixed solution to react at 90°C for 2 hours, the unreacted DMS was distilled off under reduced pressure to yield a the reactive silicon group-containing polyoxypropylene polymer. On the basis of  $^1\text{H}$ -NMR analysis, it was revealed that the introduction proportion of the the reactive silicon groups into the terminals of the obtained polymer was 77% (Polymer A). The number average molecular weight ( $M_n$ ) and the  $M_w/M_n$  value of the obtained polymer were about 15,000 and 1.1, respectively.

[0059]

(Preparation Example 2)

Propylene oxide was polymerized by use of a polyoxypropylene diol (Actocole P-23 manufactured by Mitsui Takeda Chemicals, Inc.) having a number average molecular weight of 4000 as measured by GPC as initiator and a zinc hexacyanocobaltate complex as catalyst to yield a polyoxypropylene diol having a number average molecular weight of 30000 as measured by GPC. To 1000 g of the obtained polyoxypropylene diol, 19.7 g of a 28% methanol solution of sodium methylate was added, and thereafter the methanol was distilled off at 130°C under stirring under a reduced pressure generated by a vacuum pump. Then, 9.4 g of allyl chloride was added, and the reaction mixture was stirred for 30 minutes. Then, the remaining ally chloride was removed under a reduced pressure generated by a vacuum pump. To the reaction mixture, 9.6 g of a 28% methanol solution of sodium methylate was added, and then the methanol was distilled off at 130°C under stirring under a reduced pressure generated by a vacuum pump. Subsequently, 6.3 g of ally chloride was added, the reaction mixture thus obtained was stirred for 4 hours, and then the remaining ally chloride was removed under a reduced pressure generated by a vacuum pump, to yield a crude, unsaturated group-terminated organic polymer. The obtained organic polymer was dissolved in hexane, then the hexane solution was mixed with water to undergo desalting purification, and then water was removed and hexane was evaporated to yield a pale yellow, transparent unsaturated group-terminated organic polymer. To the obtained pale yellow, transparent unsaturated group-terminated organic polymer, 10.2 g of methyldimethoxysilane and 0.06 g of an isopropanol solution of chloroplatinic acid was added as catalyst, and the reaction mixture thus obtained was stirred at 90°C for 2 hours, to yield a hydrolyzable silicon group-terminated organic polymer. The number average molecular weight ( $M_n$ ) and the  $M_w/M_n$  value

of the obtained polymer were about 30000 and 1.2, respectively, as measured by GPC.

【0060】

(Synthesis Examples 1 to 9)

In each of Synthesis Examples 1 to 9, a toluene solution in which monomers (each having a polymerizable vinyl group) shown in Table 1 and 2,2'-azobis(2-methylbutylonitrile) (trade name: V-59, manufactured by Wako Pure Chemical Industries, Ltd.) (polymerization initiator) were dissolved in the amounts respectively specified in Table 1 was added dropwise to toluene heated to 105°C over a period of 4 hours, and then polymerization was further allowed to proceed for 2 hours; thus the synthesis products 1 to 9 were obtained. The used amounts of the respective ingredients shown in Table 1 are given in parts by weight. The molecular weights of the obtained synthesis products were measured by GPC.

【0061】

(Synthesis Examples 10 to 17)

In each of Synthesis Examples 10 to 17, a toluene solution in which monomers (each having a polymerizable vinyl group) shown in Table 2 and 2,2'-azobis(2-methylbutylonitrile) (trade name: V-59, manufactured by Wako Pure Chemical Industries, Ltd.) (polymerization initiator) were dissolved in the amounts respectively specified in Table 2 was added dropwise to toluene heated to 105°C over a period of 4 hours, and then polymerization was further allowed to proceed for 2 hours; thus the synthesis products 10 to 17 were obtained. The used amounts of the respective ingredients shown in Table 2 are given in parts by weight. The molecular weights of the obtained synthesis products were measured by GPC.

[0062]

(Examples 1 to 8)

The synthesis products 1 to 8 respectively obtained in above Synthesis Examples 1 to 8 each were mixed with Polymer A obtained in Preparation Example 1 in a solid content weight ratio of 40:60, and then the toluene was distilled off at 120°C under reduced pressure, to yield the polymers of Examples 1 to 8.

[0063]

(Example 9)

The synthesis product 9 obtained in above Synthesis Example 9 was mixed with Polymer B obtained in Preparation Example 2 in a solid content weight ratio of 40:60, and then the toluene was distilled off at 120°C under reduced pressure, to yield the polymer of Example 9.

[0064]

(Comparative Examples 1 to 8)

The synthesis products 1 to 8 respectively obtained in above Synthesis Examples 10 to 17 each were mixed with Polymer A obtained in Preparation Example 1 in a solid content weight ratio of 40:60, and then the toluene was distilled off at 120°C under reduced pressure, to yield the polymers of Comparative Examples 1 to 8.

[0065]

The polymers thus obtained were allowed to stand under a condition of 5°C for 30 days, and then the conditions of each of the polymers were visually checked. The obtained polymers satisfactory in storage stability were mixed with the following ingredients to yield curable compositions: 150 parts by weight of a surface treated calcium carbonate (Hakuenka CCR: Shiraishi Kogyo), 20 parts by weight of ground calcium carbonate (Nanox 25A: Maruo Calcium), 10 parts by weight of titanium

oxide (Tipaque R-820: Ishihara Sangyo), 60 parts by weight of PPG (polypropylene glycol) having a number average molecular weight of 3,000 as plasticizer, 2 parts by weight of an antisagging agent (Disparlon #6500: Kusumoto Chemicals), 1 part by weight of an ultraviolet absorber (Tinuvin 213: Ciba Specialty Chemicals), 1 part by weight of photostabilizer (Sanol LS770: Sankyo), 2 parts by weight of a dehydrating agent (A-171: Japan Unicar), 2 parts by weight of an adhesion-imparting agent (A-1120: Japan Unicar), and 2 parts by weight of a curing catalyst (Neostann U-220: Nitto Kasei). The obtained curable compositions each were spread into a 3 mm thick sheet, and the sheets thus obtained were allowed to stand at 23°C for 3 days, and thereafter heated at 50°C for 4 days to yield blended material sheets, which were subjected to weather resistance evaluation by means of a sunshine weatherometer. The results obtained are shown in Tables 3.

[0066]

Here, it is to be noted that the abbreviations for the monomers in Tables 1 and 2 are as follows:

MMA: Methyl methacrylate

BA: Butyl acrylate

2-EHA: 2-Ethylhexyl acrylate

ANO: n-Octyl acrylate

AIO: Isooctyl acrylate

AIN: Isononyl acrylate

SMA: Stearyl methacrylate

LMA: Lauryl methacrylate

[0067]

[Table 1]

		No. of C atoms in alkyl chain	Condition of monomer	Syn. Ex. 1	Syn. Ex. 2	Syn. Ex. 3	Syn. Ex. 4	Syn. Ex. 5	Syn. Ex. 6	Syn. Ex. 7	Syn. Ex. 8	Syn. Ex. 9
Acrylic monomers	MMA	1	liquid	15	45	15	15	15	15	15	15	15
	EA	4	liquid	40	25	40	40	40	40	40	41.5	41.5
	2-EHA	8	liquid	40	25							
	ANO	8	liquid			40						41.5
	AIO	8	liquid				40					
	AIN	9	liquid					40	40	40	41.5	
	SMA	18	solid									
	LMA	12	solid									
	Y-9910	-	liquid	5	5	5	5	5	5		2	2
	A-174	-	liquid							5		
Polymerization initiator	V-59	-	-	2.0	2.0	2.0	2.0	2.0	2.0	1.4	0.5	0.5
Mol. wt	GPC-Mn	-	-	8800	9500	8700	9000	8900	8900	13000	19400	19200



[0068]

[Table 2]

		No. of C atoms in alkyl Chain	Condition of monomer	Syn. Ex. 10	Syn. Ex. 11	Syn. Ex. 12	Syn. Ex. 13	Syn. Ex. 14	Syn. Ex. 15	Syn. Ex. 16	Syn. Ex. 17
Acrylic monomers	MMA	1	liquid		20	20		15	15	15	15
	BA	4	liquid	47.5		80	20	60	60	60	20
	2-EHA	8	liquid	47.5	80		80			20	60
	ANO	8	liquid								
	AIO	8	liquid								
	AIN	9	liquid								
	SMA	18	solid					20			
	IMA	12	solid						20		
	Reactive silyl group	-	liquid	5	5	5	5	5	5	5	5
	containing monomers	-	liquid								
Polymerization initiator	V-59	-	-	2.0	2.0	2.0	2.0	2.0	2.0	2.0	2.0
Mol. wt	GPC-Mn	-	-	8800	9000	9400	9000	9200	8900	8800	8800

[0069]

[Table 3]

	Ex. 1	Ex. 2	Ex. 3	Ex. 4	Ex. 5	Ex. 6	Ex. 7	Ex. 8	Ex. 9
Polym. 1	40								
Polym. 2		40							
Polym. 3			40						
Polym. 4				40					
Polym. 5					40				
Polym. 6						40			
Polym. 7							40		
Polym. 8								40	
Polym. 9									40
Reactive silyl group terminated polypropylene oxide	60	60	60	60	60	60	60	60	60
Storage stability *1	O	O	O	O	O	O	O	O	O
Weather resistance *2	O	O	O	O	O	O	O	O	O
Polym. 10	40								
Polym. 11		40							
Polym. 12			40						
Polym. 13				40					
Polym. 14					40				
Polym. 15						40			
Polym. 16							40		
Polym. 17								40	
Reactive silyl group terminated polypropylene oxide	60	60	60	60		60	60	60	100
Storage stability *1	X	X	X	X	O	O	X	X	O
Weather resistance *2	-	-	-	-	O	O	-	-	X

\*1: O Transparent liquid; X Precipitation.

\*2: O No cracking after 2000 hours; X Cracking after 2000 hours.

【0070】

【EFFECT OF THE INVENTION】

As is clear from table 3, the reactive modifier of examples 1 to 9 of the present invention is inexpensive in the raw materials used, easy to handle, and has the same storage stability as conventional one.

【Document Name】 ABSTRACT

【Abstract】

【Subject】

The present invention provides a reactive modifier which is inexpensive in the raw materials thereof, easy to handle and excellent in the storage stability as observed at the time of mixing with a polymer.

【Solution means】

A reactive modifier which is an acrylic polymer having a number averagemolecular weight of 5000 or more, obtained by copolymerizing (A) a hydrolyzable silyl group-containing monomer (B) methyl methacrylate, (C) butyl acrylate and (D) an alkyl (meth) acrylate in which the alkyl chain has 7 to 9 carbon atoms, wherein:

the content of the component A is 0.01 to 10 parts by weight, the content of the component B is 5 to 95 parts by weight, the content of the component C is 5 to 95 parts by weight, the content of the component D is 5 to 95 parts by weight, the total content of the components A to D is 50 to 100 parts by weight and the weight ratio of the component C to the component D is 0.5 to 2.0.

【Figure selected】 none